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#### US – Late 1980s

- Researchers note that volatile organic compounds (VOCs) are the most common groundwater constituent at landfills
- US landfill groundwater monitoring requirements modified to include VOCs as 'indicator parameters', due to frequency and mobility
- VOCs in groundwater from landfill gas (LFG) had not been envisioned
- Due to much lower cost for gas control than leachate control, differentiating gas from leachate important.

- Case Study 1 Use of radioisotopes to assess LFG impacts on groundwater
   (http://www.ncbi.nlm.nih.gov/pubmed/14710929) and inorganic effects of LFG on groundwater
   (http://info.ngwa.org/gwol/pdf/040278119.pdf)
- Case Study 2 Comparison of gas and aqueous data to assess the direction of partitioning in a monitoring well; In-well LFG effects

(http://www.ncbi.nlm.nih.gov/pubmed/15736343)

- Within-well vs Out-of-well LFG impacts
- Time to cleanup

- Case Study 1 Use of radioisotopes to assess LFG impacts and inorganic effects of LFG on groundwater
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### Case Study 1: MSW Landfill: LA Area

- Spreading grounds (MAR) upgradient
- Quarry downgradient
- Partially lined, partially unlined
- Persistent low-level VOC detections DCDFM, TCE upgradient and cross-gradient...
- Others -- GW and gas VOCs 'at equilibrium'
- Permit renewal at stake

#### **Isotope evaluation performed:**

- A. Dissolved inorganic carbon (Carbonates) from LFG CO<sub>2</sub> or leachate alkalinity -- modern, high <sup>14</sup>C level vs. background carbonates
- B. Leachate water likely has elevated tritium (<sup>3</sup>H)

Elevated <sup>14</sup>C and tritium -> Leachate;

Elevated <sup>14</sup>C only -> LFG







Evaluated changes in inorganic constituents during and after LFG effects on groundwater



#### **Parameters Evaluated:**

- Alkalinity
- TDS
- Fe and Mn
- Na, SO4 and Cl
- Ca and Mg









# **Calcium and Magnesium and Alkalinity**



### LFG Can Cause Changes in Non-Volatile Constituents

- Alkalinity
- Ca and Mg
- Manganese
- No Effect on Sodium, Chloride

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- Operating unlined landfill
- New low-permeability cap and landfill gas extraction system (LFGCCS) installed
- Before LFGCCS operation began, detections of volatile organic compounds cross-gradient and upgradient from waste
- Regulators required explanation for detections

#### Volatile Organic Compounds Detected In Ground-Water Samples

Compound	Henry's Law Coefficient	OW-08 (ug/L)	OW-10 (ug/L)	OW-13 (ug/L)	OW-15 (ug/L)
<u>cis</u> -1,2-Dichlchloroethene	0.31	ND(10) <sup>a</sup>	10	ND(10)	10
1,1-Dichloroethane	0.23	ND(5)*	7	ND(5)	10
Dichloromethane	0.11	ND(5)	ND(5)	ND(5)	8
Trichloroethene	0.37	ND(1.4)	2	ND(1.4)	3
Tetrachloroethene	0.93	ND(1.1)	3	ND(1.1)	5

## **In-Well Gas Effects**



## **In-Well Gas Effects**





Evaluating gas/water equilibrium by comparing gas and aqueous concentrations

EQUILIBRIUM (No net phase transfer):

 $C_g = H C_w$ 

Gas-to-Water Transfer:

$$C_g > H C_w$$

Water-to-Gas Transfer:

$$C_g < H C_w$$

#### Table 3. Gas Concentrations(ng/cm<sup>3</sup>) and Evaluation of Gas/Water Equilibrium of Volatile Organic Compounds<sup>a</sup>

Well	<u>cis</u> -1,2-DCE (H=0.31)		1,1 (H=	-DCA =0.23)	MeCl <sub>2</sub> (H=0.11)		TCE (H=0.37)		PCE (H=0.93)	
		HC C <sub>s</sub>		HC Cg		HC C <sub>s</sub>		HC C <sub>g</sub>		HC C
OW-08		NW <sup>c</sup>		NW <sup>c</sup>		NG⁴		NW <sup>c</sup>		NW <sup>c</sup>
OW-10		0.03		NG <sup>d</sup>		NG		0.8		0.19
OW-13		NG		NG <sup>d</sup>		NG		NG⁴		NA
OW-15		0.32		0.77		0.35		0.41		1.4

<sup>a</sup> Gas concentrations (C<sub>g</sub>) in ng/cm<sup>3</sup>; HC<sub>w</sub>/C<sub>g</sub> dimensionless <sup>b</sup> Non-detect; See Methods and Materials for detection limits

<sup>c</sup> Entries of NW represent non-detectable water concentrations with detectable gas concentrations

<sup>d</sup> Entries of NG represent non-detectable gas concentrations



## **In-Well Gas Effects**

Compound	OW-08	OW-10	OW-13	OW-15
Dichlorodifluoromethane	1.2	1.4	ND(0.003)*	ND(0.002)*
1,2-Dichloro-1,1,2,2- Tetra fluoroethane	0.42	ND(0.0009)	ND(0.007)*	0.29
Vinyl Chloride	1.9	0.69	ND(0.002)*	2.3
Chloroethane	1.6	ND(0.0008)*	ND(0.003)*	3.2
Trichlorofluoromethane	0.007	0.31	ND(0.002) <sup>a</sup>	0.90
1,1,2-Trichloro-1,2,2- trifluoroethane	0.09	ND(0.001)*	ND(0.005)*	0.23
trans-1,2-Dichloroethene	ND(0.005)*	ND(0.001)*	ND(0.005)*	0.17
1,1-Dichloroethene	ND(0.003)*	ND(0.005)*	ND(0.003)*	0.10
Chloroform	ND(0.003)*	ND(0.0006)*	ND(0.003) <sup>a</sup>	0.12
Benzene	1.2	0.35	ND(0.003)*	2.0
1,2-Dichloroethane	ND(0.003)*	ND(0.005)*	ND(0.003) <sup>a</sup>	0.09
Toluene	0.11	1.0	0.079	3.4
Chlorobenzene	. ND(0.004)*	ND(0.008)*	ND(0.004) <sup>a</sup>	0.12
Ethylbenzene	ND(0.003)*	0.35	ND(0.004) <sup>a</sup>	2.1
Total Xylenes	1.1	0.61	0.065	7.8

#### Table 4.Concentrations (ng/cm³) of Compounds Detected in Monitoring-WellHeadspace Gases but Absent from all Ground-Water Samples

<sup>a</sup>ND = Not Detected; Detection limit in ng/cm<sup>3</sup> in parentheses.

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# **Timeframe of Corrective Actions for LFG Effects**

## **In-Well Effects**

 In-well LFG effects: When gas/water contact is limited to within the bore

## **Out-of-Well Effects**

 Out-of-well LFG effects: When gas/water contact occurs outside (upgradient) of the bore

# **Timeframe of Corrective Actions for LFG Effects**

#### **In-Well Effects**

- Can be due to well construction, stratigraphy -- Gas/water contact area (in well) known
- Corrective action can potentially be targeted to intercept transport pathway
- Due to limited gas/water contact area, MNA possible

#### **Out-of-Well Effects**

- A potentially dispersed and unknown gas/water contact area (possibly beneath waste)
- Enhanced landfill gas collection (source control) best approach

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# Time to Re-Establish Compliance Influenced by:

- GW Concentrations;
- GW Flow Velocity;
- Vadose Zone Properties;
- Residuals of LFG; and
- Others



- Very Rapid GW Flow
- Deep Vadose Zone
- Highly Variable Concentrations During Time of LFG Effects
- Approximately 48 Months for Recovery to ND in one case

## **In-Well Gas Effects**







Thank you

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